Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

Catharine Esterhuysen,\* Julia Even, Matthias W. Esterhuysen, Ulrike E. I. Horvath, Elzet Stander and Helgard G. Raubenheimer

Department of Chemistry and Polymer Science, University of Stellenbosch, Private Bag X1, Matieland 7602, South Africa

Correspondence e-mail: ce@sun.ac.za

#### **Key indicators**

Single-crystal X-ray study T = 173 K Mean  $\sigma$ (C–C) = 0.003 Å R factor = 0.032 wR factor = 0.077 Data-to-parameter ratio = 17.1

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

# Pentacarbonyl[methyl(*n*-propylsulfanyl)carbene]chromium(0)

The title compound,  $[Cr(C_5H_{10}S)(CO)_5]$ , comprises a methyl(*n*-propylsulfanyl)carbene ligand coordinated to a pentacarbonylchromium fragment, with the Cr atom in an octahedral coordination. Received 9 June 2006 Accepted 14 June 2006

## Comment

In our ongoing study of the chemistry of Fischer-type carbene complexes, we have focused on the utilization of simple derivatives as organometallic synthons (Raubenheimer *et al.*, 2002, 2005). In a recent report (Raubenheimer *et al.*, 2006), we discussed the reaction of deprotonated thiocarbene complexes with the electrophile  $Ph_3PAu^+$  to afford mercaptovinyl complexes of Au that are S-coordinated to the  $M(CO)_5$  (M = Cr, W) unit of the original Fischer carbene complex. Since structural information on the starting thiocarbene complexes is limited, the structure of the title compound, (I), is reported here.



The molecular structure of (I) (Fig. 1) comprises a methyl(*n*-propylsulfanyl)carbene ligand coordinated to a central Cr atom. Five carbonyl ligands complete the octahedral coordination around chromium. The thiocarbene is staggered with respect to the equatorial CO ligands, with S-C-Cr-Ctorsion angles of 39.70 (12), 129.69 (11), -140.28 (11) and -49.06 (11)°. The two S-C bond lengths (Table 1) differ significantly, suggesting that a certain amount of  $\pi$ -donation from the heteroatom to the carbene C atom occurs, thus shortening the C11-S11 bond. This is similar to the structure of pentacarbonyl[methyl(phenylthio)carbene]chromium (Hoare & Mills, 1972), where the S-C(carbene) and S-C(phenyl) distances are 1.690 (3) and 1.781 (3) Å, respectively.

# Experimental

Compound (I) was prepared according to a somewhat modified version of a literature procedure (Aumann & Schroeder, 1990). Dropwise addition of 1-propanethiol (0.28 ml, 3.09 mmol) to a yellow methanolic solution (20 ml) of pentacarbonyl[methoxy(meth-yl)carbene]chromium(0) (0.76 g, 3.04 mmol) and Na<sub>2</sub>CO<sub>3</sub> (0.32 g, 3.02 mmol), cooled to 268 K, afforded a dark-red solution. After 15 min at 273 K, extraction with a mixture of *n*-hexane/water (200/ 10 ml) and drying of the organic layer over Na<sub>2</sub>SO<sub>4</sub>, followed by final

© 2006 International Union of Crystallography All rights reserved

# metal-organic papers

stripping of solvent, yielded a microcrystalline product. Recrystallization from *n*-hexane at 195 K provided dark-red needles (0.55 g, 62%; m.p. 316 K). IR [*n*-hexane,  $\nu$ (CO) cm<sup>-1</sup>]: 2059 (A<sub>1</sub>), 1988 (B<sub>1</sub>), 1957 (E); <sup>1</sup>H NMR (CD<sub>3</sub>COCD<sub>3</sub>):  $\delta$  3.71 (*s*, 3 H, CH<sub>3</sub>), 3.40 (*t*, 2 H, *J* = 7.5 Hz, SCH<sub>2</sub>), 1.77 (*m*, 2 H, *J* = 7.5 Hz, SCH<sub>2</sub>-CH<sub>2</sub>), 1.04 [*t*, 3 H, *J* = 7.5 Hz, S(CH<sub>2</sub>)<sub>2</sub>-CH<sub>3</sub>]; <sup>13</sup>C NMR (CD<sub>3</sub>COCD<sub>3</sub>):  $\delta$  369.8 (C<sub>carb</sub>), 228.5 (CO<sub>trans</sub>), 217.9 (CO<sub>cis</sub>), 46.9 (CH<sub>3</sub>), 45.5 (SCH<sub>2</sub>), 21.3 (SCH<sub>2</sub>-CH<sub>2</sub>), 13.3 [S(CH<sub>2</sub>)<sub>2</sub>-CH<sub>3</sub>]; EI-MS (*m*/*z*): 294 [*M*]<sup>+</sup>, 238 [*M* - 2CO]<sup>+</sup>, 182 [*M* - 4CO]<sup>+</sup>, 154 [*M* - 5CO]<sup>+</sup>, 112 [*M* - C<sub>3</sub>H<sub>7</sub>]<sup>+</sup>.

### Crystal data

$[Cr(C_5H_{10}S)(CO)_5]$	$V = 632.17 (12) \text{ Å}^3$
$M_r = 294.24$	Z = 2
Triclinic, P1	$D_x = 1.546 \text{ Mg m}^{-3}$
a = 6.1320 (7) Å	Mo $K\alpha$ radiation
b = 8.4685 (9)  Å	$\mu = 1.07 \text{ mm}^{-1}$
c = 12.6187 (14)  Å	T = 173 (2) K
$\alpha = 77.264 \ (2)^{\circ}$	Needle, dark red
$\beta = 81.530 \ (2)^{\circ}$	$0.30 \times 0.10 \times 0.10$ mm
$\gamma = 87.878 \ (2)^{\circ}$	

### Data collection

Bruker APEX CCD area-detector diffractometer  $\omega$  scans Absorption correction: multi-scan (*SADABS*; Bruker, 2001)  $T_{\min} = 0.764, T_{\max} = 0.900$ 

## Refinement

Refinement on $F^2$	$w = 1/[\sigma^2(F_0^2) + (0.0352P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.032$	+ 0.4906P]
$wR(F^2) = 0.077$	where $P = (F_0^2 + 2F_c^2)/3$
S = 1.07	$(\Delta/\sigma)_{\rm max} = 0.001$
2661 reflections	$\Delta \rho_{\rm max} = 0.42 \text{ e } \text{\AA}^{-3}$
156 parameters	$\Delta \rho_{\rm min} = -0.25 \text{ e } \text{\AA}^{-3}$
H-atom parameters constrained	

6077 measured reflections

 $R_{\rm int} = 0.017$ 

 $\theta_{\rm max} = 26.8^\circ$ 

2661 independent reflections

2434 reflections with  $I > 2\sigma(I)$ 

#### Table 1

Selected	geometric	parameters	(Å, °	).
----------	-----------	------------	-------	----

Cr1-C11	2.0442 (19)	\$11-C13	1.8229 (19)
\$11-C11	1.6802 (19)	C11-C12	1.499 (3)
C11-S11-C13	112.20 (9)	Cr1-C11-S11	117.27 (10)



#### Figure 1

The molecular structure and atom-labelling scheme for (I), showing 50% probability displacement ellipsoids.

The H atoms were included in the riding-model approximation, with methylene C-H = 0.99 Å and methyl C-H = 0.98 Å, and with  $U_{\rm iso}({\rm H}) = 1.2$  and  $1.5U_{\rm eq}({\rm C})$  for methylene and methyl H atoms, respectively.

Data collection: *SMART* (Bruker, 2001); cell refinement: *SAINT* (Bruker, 2002); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP* for Windows (Farrugia, 1997); software used to prepare material for publication: *X-SEED* (Barbour, 2001).

We thank the NRF and the University of Stellenbosch for financial support.

## References

- Aumann, R. & Schroeder, J. (1990). Chem. Ber. 123, 2053-2058.
- Barbour, L. J. (2001). J. Supramol. Chem. 1, 189-191.
- Bruker (2001). SMART (Version 5.625) and SADABS (Version 2.03a). Bruker AXS Inc., Madison, Wisconsin, USA.
- Bruker (2002). SAINT (Version 6.36a). Bruker AXS Inc. Madison, Wisconsin, USA.
- Farrugia, L. J. (1997). J. Appl. Cryst. 30, 565.
- Hoare, R. J. & Mills, O. S. (1972). J. Chem. Soc. Dalton Trans. pp. 653-656.
- Raubenheimer, H. G., Esterhuysen, M. W. & Esterhuysen, C. (2005). Inorg. Chim. Acta, 358, 4217–4228.
- Raubenheimer, H. G., Esterhuysen, M. W., Frenking, G., Timoshkin, A., Esterhuysen, C. & Horvath, U. E. I. (2006). *J. Chem. Soc. Dalton Trans.* Accepted.
- Raubenheimer, H. G., Esterhuysen, M. W., Timoshkin, A., Frenking, G. & Chen, Y. (2002). Organometallics, 21, 3173–3181.
- Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. Versions 97-1. University of Göttingen, Germany.